

IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet PCI20

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$\text{CH}_3\text{CCl}_3 + h\nu \rightarrow \text{products}$

Primary photochemical processes

| Reaction | $\Delta H^\circ/\text{kJ}\cdot\text{mol}^{-1}$ | $\lambda_{\text{threshold}}/\text{nm}$ |
|--|--|--|
| $\text{CH}_3\text{CCl}_3 + h\nu \rightarrow \text{CH}_3\text{CCl}_2 + \text{Cl}$ | 335 (est) | 360 |

Preferred Values

Absorption cross-sections for CH_3CCl_3 at 295 K and 210 K

| λ/nm | $10^{20} \sigma/\text{cm}^2$ | | λ/nm | $10^{20} \sigma/\text{cm}^2$ | |
|---------------------|------------------------------|-------|---------------------|------------------------------|-------|
| | 295 K | 210 K | | 295 K | 210 K |
| 182 | 315 | a | 210 | 24.0 | 19.8 |
| 184 | 280 | a | 212 | 16.8 | 13.2 |
| 186 | 250 | a | 214 | 12.0 | 8.8 |
| 188 | 220 | a | 216 | 8.6 | 6.1 |
| 190 | 192 | a | 218 | 6.0 | 4.2 |
| 192 | 163 | a | 220 | 4.1 | 2.9 |
| 194 | 140 | a | 222 | 2.9 | 1.2 |
| 196 | 118 | a | 224 | 2.0 | 1.2 |
| 198 | 99 | a | 226 | 1.5 | 0.76 |
| 200 | 81 | a | 228 | 1.0 | 0.51 |
| 202 | 66 | 64 | 230 | 0.70 | 0.33 |
| 204 | 52 | 49 | 232 | 0.49 | 0.18 |
| 206 | 40 | 36 | 234 | 0.33 | 0.11 |
| 208 | 31 | 26 | 236 | 0.23 | 0.064 |
| | | | 238 | 0.15 | 0.036 |
| | | | 240 | 0.10 | 0.024 |

(a) No temperature dependence observed.

Comments on Preferred Values

The preferred values of the absorption cross-sections at 298 K and at 210 K are the values reported by Vanlaethem-Meuree et al. (1979), who determined absorption cross-section values at 295 K, 270 K, 250 K, 230 K and 210 K for the wavelength range 180–240 nm. These values (Vanlaethem-Meuree et al., 1979) are preferred over the substantially higher values reported by Hubrich and Stuhl (1980), in which study a correction was required for the presence of the UVabsorbing stabilizer 1,4-dioxane. In a recent study, Nayak et al. (1995) reported measurements in the gas phase (160–240 nm) and the liquid phase (235–260 nm) over the temperature range 220–330 K.

A wavelength shift procedure was used to convert the liquid-phase values into effective gas-phase values at the long wavelengths. The reported room temperature values of Nayak et al. (1995) are in good agreement (within 15%) with those of Vanlaethem-Meuree et al. (1979) in the range 210–240 nm, whereas in the 180–210 nm range they are 15% to 30% higher. Low temperature results are in relatively good agreement up to 230 nm. Photolysis is expected to occur with unit quantum efficiency by breaking of the C-Cl bond to yield $\text{CH}_3\text{CCl}_2 + \text{Cl}$.

References

Hubrich, C. and Stuhl, F.: *J. Photochem.* 12, 93, 1980.

Nayak, A. K., Kurylo, M. J. and Fahr, A.: *J. Geophys. Res.* 100, 11185, 1995.

Vanlaethem-Meuree, N., Wisenberg, J. and Simon, P. C.: *Geophys. Res. Lett.* 6, 451, 1979.