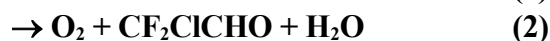
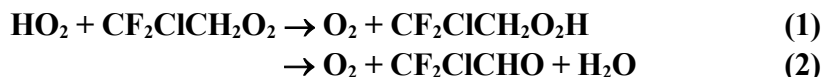


# IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet oClOx77

Website: <http://iupac.pole-ether.fr>. See website for latest evaluated data. Data sheets can be downloaded for personal use only and must not be retransmitted or disseminated either electronically or in hardcopy without explicit written permission.

This data sheet last evaluated: June 2015; last change in preferred values: December 2004.



## Rate coefficient data ( $k = k_1 + k_2$ )

$k/\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$	Temp./K	Reference	Technique/ Comments
<i>Absolute Rate Coefficients</i>			
$(6.8 \pm 3.4) \times 10^{-12}$	296	Hayman and Battin-Leclerc, 1995	LP-UVA (a)

## Comments

- (a) Flash photolysis of  $\text{H}_2\text{O}_2$  in the presence of  $\text{CF}_2\text{ClCH}_3\text{-O}_2\text{-N}_2$  mixtures at a total pressure of 1013 mbar. Decays in transient absorption signals (with contributions from  $\text{CF}_2\text{ClCH}_2\text{O}_2$  and  $\text{HO}_2$ ) were recorded in the wavelength range 220 nm to 240 nm.  $k$  derived from simulations of the decay traces using a 10 reaction mechanism.

## Preferred Values

$$k = 6.8 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1} \text{ at } 298 \text{ K.}$$

### Reliability

$$\Delta \log k = \pm 0.5 \text{ at } 298 \text{ K.}$$

### Comments on Preferred Values

While the above value of the rate coefficient seems reasonable, it has been derived from the analysis of a complex chemical system and requires independent verification to reduce the recommended error limits. Within the uncertainty of the determination,  $k$  is indistinguishable from that recommended for the reaction of  $\text{HO}_2$  with  $\text{C}_2\text{H}_5\text{O}_2$ , suggesting that the presence of the  $\text{CF}_2\text{Cl}$  group has only a minor influence on the reaction.

## References

Hayman, G. and Battin-Leclerc, F.: J. Chem. Soc. Farad. Trans. 91, 1313, 1995.