IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet R Oxygen 6

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This data sheet updated: 12th June 2003.

$$i-C_3H_7 + O_2 + M \rightarrow i-C_3H_7O_2 + M$$

 $\Delta H^{\circ} = -158.9 \text{ kJ} \cdot \text{mol}^{-1}$

High-pressure rate coefficients

Rate coefficient data

k∞/cm³ molecule⁻¹ s⁻¹	Temp./K	Reference	Technique/ Comments
Absolute Rate Coefficients			
$(1.41 \pm 0.24) \times 10^{-11}$ 8.3 x 10 ⁻¹²	298 300	Ruiz and Bayes, 1984 ¹ Munk <i>et al.</i> , 1986 ²	FP-MS (a) (b)

Comments

- (a) No pressure dependence detected for He or N₂ pressures from 1.3 mbar to 5 mbar.
- (b) Pulsed radiolysis in H₂ at 1 bar. *i*-C₃H₇ radicals were generated by the addition of H atoms to C₃H₆ and *i*-C₃H₇O₂ detected by UV absorption at 253 nm. Absorption spectrum of *i*-C₃H₇ was also detected.

Preferred Values

 $k \approx k_{\infty}$ at 298 K and 1 bar of air.

 $k_{\infty} = 1.1 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$, independent of temperature over the range 200 K to 300 K.

Reliability

 $\Delta \log k_{\infty} = \pm 0.3$ over the range 200 K to 300 K.

Comments on Preferred Values

The preferred values are the average of the results from refs. 1 and 2. Falloff corrections are probably within the uncertainties of the average. The rate coefficient k_{∞} for this reaction appears consistent with those for the reactions $C_2H_5 + O_2 + M \rightarrow C_2H_5O_2 + M$ and n- $C_3H_7 + O_2 + M \rightarrow n$ - $C_3H_7O_2 + M$ (see this evaluation).

References

- ¹ R. P. Ruiz and K. D. Bayes, J. Phys. Chem. **88**, 2592 (1984).
- ² J. Munk, P. Pagsberg, E. Ratajczak, and A. Sillesen, Chem. Phys. Lett. **132**, 417 (1986).