IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet Ox5

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$$O_2* + O_3 \rightarrow O + 2O_2$$

Comments

Arnold and Comes^{1,2} have studied this reaction of vibrationally excited oxygen molecules in the ground electronic state with ozone and the yreport a rate coefficient of 2.8×10^{-15} cm³ molecule⁻¹ s¹ at 298 K. The vibrationally excited oxygen molecules were produced in the reaction of $O(^1D)$ atoms with O_3 following the UV photolysis of ozone. This is the only reported study of this rate coefficient, and we make no recommendation. For further discussion the reader is referred to the review by Steinfeld *et al.*³

References

- ¹ I. Arnold and F. J. Comes, Chem. Phys. **47**, 125 (1980).
- ² I. Arnold and F. J. Comes, J. Mol. Struct. **61**, 223 (1980).
- ³ J. I. Steinfeld, S. M. Adler-Golden, and J. W. Gallagher, J. Phys. Chem. Ref. Data **16**, 911 (1987).