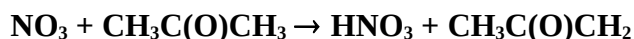


## IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet NO<sub>3</sub>\_VOC13

Website: <http://iupac.pole-ether.fr>. See website for latest evaluated data. Data sheets can be downloaded for personal use only and must not be retransmitted or disseminated either electronically or in hardcopy without explicit written permission.

This data sheet updated: 12<sup>th</sup> December 2007 (with no revision of the preferred values).



$$\Delta H^\circ = -15.5 \text{ kJ}\cdot\text{mol}^{-1}$$

### Rate coefficient data

$k/\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$	Temp./K	Reference	Technique/ Comments
<i>Absolute Rate Coefficients</i> $\leq (8.5 \pm 2.5) \times 10^{-18}$	302	Boyd et al., 1991	(a)

### Comments

- (a) Stopped flow system with detection of the NO<sub>3</sub> radical by optical absorption at 662 nm. Secondary reactions were believed to be important and a stoichiometry factor of  $\geq 2$  has been used to obtain the cited upper limit to the rate coefficient.

### Preferred Values

$$k < 3 \times 10^{-17} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1} \text{ at } 298 \text{ K.}$$

#### *Comments on Preferred Values*

The upper limit to the preferred value is derived from the overall rate coefficient of  $(1.7 \pm 0.5) \times 10^{-17} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  measured by Boyd et al. (1991), with no account taken of the expected greater than unity stoichiometry.

### References

Boyd, A., Canosa-Mas, C. E., King, A. D., Wayne, R. P. and Wilson, M. R.: J. Chem. Soc. Faraday Trans., 87, 2913, 1991.