

## IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet HO<sub>x</sub>\_VOC47

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This data sheet last evaluated: 2<sup>nd</sup> August 2007; no revision of preferred values.

### HO + CH<sub>3</sub>CH(ONO<sub>2</sub>)C(O)CH<sub>3</sub> → products

#### Rate coefficient data

$k/\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$	Temp./K	Reference	Technique/ Comments
<i>Relative Rate Coefficients</i> (1.15 ± 0.13) × 10 <sup>-12</sup>	298 ± 2	Zhu et al., 1991	RR (a)

#### Comments

- (a) HO radicals were generated by the photolysis of CH<sub>3</sub>ONO-NO-N<sub>2</sub>-O<sub>2</sub> mixtures at 1 bar pressure. The concentrations of CH<sub>3</sub>CH(ONO<sub>2</sub>)C(O)CH<sub>3</sub> and *n*-butane were measured during the experiments by GC, and the measured rate constant ratio of  $k(\text{HO} + \text{CH}_3\text{CH}(\text{ONO}_2)\text{C}(\text{O})\text{CH}_3)/k(\text{HO} + n\text{-butane}) = 0.499 \pm 0.056$  is placed on an absolute basis by use of a rate coefficient of  $k(\text{HO} + n\text{-butane}) = 2.3 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  (IUPAC, current recommendation).

#### Preferred Values

$$k = 1.2 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1} \text{ at } 298 \text{ K.}$$

#### Reliability

$$\Delta \log k = \pm 0.3 \text{ at } 298 \text{ K.}$$

#### Comments on Preferred Values

The preferred value is based on the sole study of Zhu et al. (1991), but with a higher uncertainty.

#### References

IUPAC,; <http://iupac.pole-ether.fr>, 2013.

Zhu, T., Barnes, I. and Becker, K. H.: J. Atmos. Chem. 13, 301, 1991.