

## IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet CGI\_8

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This datasheet updated: May 2015; .

### CH<sub>2</sub>OO + CH<sub>3</sub>(CO)CH<sub>3</sub> → products

#### Rate coefficient data

$k/\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$	Temp /K	Reference	Technique/Comments
Absolute Rate Coefficients			
$(2.3 \pm 0.3) \times 10^{-13}$	293	Taatjes et al., 2012	PLP-PIMS(a)

#### Comments

(a) CH<sub>2</sub>OO (formaldehyde oxide) was produced by the reaction of CH<sub>2</sub>I + O<sub>2</sub>. CH<sub>2</sub>I was generated by 248-nm laser photolysis of diiodomethane, CH<sub>2</sub>I<sub>2</sub>, at 293 K and 4 Torr total pressure in a large excess of O<sub>2</sub>. The reacting mixture was monitored by tunable synchrotron photoionization mass spectrometry, which allowed characterisation of the PIMS for CH<sub>2</sub>OO and its reaction products over the region 9.5 – 11.5 eV, and time-resolved direct detection of CH<sub>2</sub>OO at  $m/z = 46$  amu. The first order decay CH<sub>2</sub>OO in the presence of excess known concentrations of acetone was used to determine the rate constants. The uncertainty limits are 95%, based on unweighted linear fit of [CH<sub>3</sub>(CO)CH<sub>3</sub>] dependence of decay lifetimes. Secondary ozonide (3,3-dimethyl-1,2,4-trioxalane) was identified as a reaction product from its PIMS aided by quantum chemical calculations.

#### Preferred Values

Parameter	Value	T/K
$k/\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$	$2.3 \times 10^{-13}$	298
<i>Reliability</i>		
$\Delta \log k$	$\pm 0.3$	298

#### Comments on Preferred Values

The direct photoionization mass spectrometric detection of formaldehyde oxide (CH<sub>2</sub>OO) as a product of the photolysis of CH<sub>2</sub>I<sub>2</sub> in the presence of O<sub>2</sub> (Welz et al., 2012) has allowed detailed investigation of the kinetics of the Criegee intermediates, including the reaction: CH<sub>2</sub>OO + CH<sub>3</sub>(CO)CH<sub>3</sub> → products. The rate measurements appear to be precise and consistent with the emerging reactivity pattern for CH<sub>2</sub>OO reactivity. The higher uncertainty reflects the single investigation.

### References

Taatjes, C.A., Welz, O., Eskola, A.K., Savee, J. D., Osborn, D. L., Lee, E.P.F., Dyke, J.M., Mok, D.W.K., Shallcross, D. E., and Percival, C. J., *Phys Chem.Chem.Phys.*, 14, 10391, 2012.